Quiz 3: I	ntroduction to Chemistry II	Lecture Spring 2021	Dr. Hahn (9 am MWF)	2/24/21 Wednesday	The Citadel
Name	Kery	200		test #	
	ort Answer portion of the tes NO partial credit.	t, please show work on a	ll questions for partial and t	full credit. On the mul	tiple choice questions,
MULTIP	LE CHOICE. Choose the	one alternative that b	est completes the statem	ent or answers the qu	uestion.
1) The segment of a polymo	er shown below repres	ents a 25 pts	on quiz	1)
	о о ~С-СН ₂ СН ₂ -С	-осн ₂ сн ₂ о~	(SEC are	pts) - bu ags 20pts	t quit
	A) polyamide.	B) polystyrene.	C) polyester.	D) polyethyl	ene.
2	monomers when the B) Addition polymers do not. C) Condensation polypolymers do not.	mers and condensation ne polymer is formed. s contain all of the atom rmers contain all of the	ion and condensation pole in polymers lose some ato ins of the original monome atoms of the original mo	ms from the original ers, but condensation nomers, but addition	
3) The substance with the f	ormula shown below i	s a(n)		3)
	CH3CH2CH2C	H ₂ CH ₂ C	CH ₂ CH ₂ CH ₂ COOH		
	A) alcohol.	B) saponin.	C) detergent.	(D) fatty acid	
4	- man	n sulfur in the water. use a ringing sound in t up by "hard" metal ions			4)
5	C) Soaps form micelle	d is ionic. nd has the hydrocarbo	n chain.	rue?	5)
6) The major advantage of A) are effective in hard C) are biodegradable.	d water.	BS detergents is that they B) lack phosphat D) are soil based.		6)

- (b) fill in the blank with a letter (A) peptide bond (B) C terminal end (C) N terminal end (6 pts)
 - 3. Chloroplasts in plants convert (3 pts)
 (A) Light energy into heat energy (B) light energy into electrical energy
 (C) light energy into chemical energy (D) heat energy into chemical energy
- 4. Iodine number is a measure of (3 pts)
- (A) the degree of unsaturation of a fat (B) Number of iodine atoms in a protein
- (C) the number of iodine atoms in a fat (D) number of iodine atoms in a carbohydrate

Extra Credit: (4 pts) Match the following type of linkage in proteins with the shown. (A) dispersion force (B) ionic bonds (salt bridge) (C) hydrogen bonding (D) disulfide linkage

