

Sign Name Please show work or	442) Fall 2015 Dr. Hahn MWF12pm 11/9/15M Form B Quiz V Exam #  Print Name  n all questions for partial credit even on questions which do not specify. I will only grade
1. Keeping in n aldehyde, complete you just give reaction	nind the general reaction mechanism of the addition of a nucleophile to a ketone or the following reaction mechanism. (MECHANISM means you show all intermediates. If on products, you will LOSE LOTS OF POINTS.) (10 pts)  (10) TO-CH <sub>2</sub> CH <sub>4</sub> (11) CH <sub>2</sub> CH <sub>4</sub> (12) CH <sub>2</sub> CH <sub>4</sub> (13) CH <sub>2</sub> CH <sub>4</sub> (14) CH <sub>3</sub> CH <sub>4</sub> (14) CH <sub>4</sub> (15) CH <sub>4</sub> CH <sub>4</sub> (15) CH <sub>4</sub> CH <sub>4</sub> (16) CH <sub>4</sub> CH <sub>4</sub> (17) CH <sub>5</sub> CH <sub>4</sub> (17) CH <sub>5</sub> CH <sub>4</sub> (18) CH <sub>4</sub> CH <sub>4</sub> (19) CH <sub>5</sub> CH <sub>5</sub> CH <sub>4</sub> (19) CH <sub>5</sub> CH <sub>5</sub> (19) CH <sub>5</sub> CH <sub>4</sub> (19) CH <sub>5</sub> CH <sub>5</sub> CH <sub>5</sub> (19) CH <sub>5</sub>
want counted. Writehoose, I will just gr	ictural formula of the Organic Product. Circle the number of the 5 reaction which you lite EC by the one reaction which you want counted as an extra credit problem If you do not rade the first 5 reactions and grade the 6th as extra credit. (EC worth 4 pts) (5 pts cach, 15 pts)  + C1f <sub>3</sub> CH <sub>2</sub> -O-H H <sup>4</sup> /H <sub>2</sub> O <sub>2</sub> Ch <sub>3</sub> -O-Ch <sub>4</sub> CH <sub>3</sub> O Ch <sub>4</sub> CH <sub>3</sub>
	Chy Chy Chy Chy Chy  Light H Br (large excess) -> Chy Br + Br Chy Chy Chy  H -H Cl  Aby C- G- Ut
4) 0.7 H — C—C	H CH3 Mg Br H CH4
5) CH <sub>2</sub> -CH	LIAITIA CUZ CUZ
6) CH;-C-H  7) CH;-C-H	CH3CH2-C-P-Ph3 — Cb3 C= CU2CU3

Organic II (CHEM 442) Fall 2015 Dr. Hahr		Quiz V Exam#			
Sign Name	Print Name	·			
Please show work on all questions for partial	credit even on questions which do not	t specify. I will only grade			
legible answers because if I can't read your answer, I obviously cannot grade it. NO EXCEPTIONS. (25 pts)					
,		color			
1. Keeping in mind the general reaction mechanism of the addition of a nucleophile to a ketone or					
aldobade complete the following reaction managed	echanism - (MECHANISM means voi	u show all intermediates. If			
you just give reaction products, you will LOS	SE LOTS OF POINTS.) (10 pts)	do not do 120)			
$CH_3 - C - H$ (a) $Na^{\dagger} C \equiv N$		add to EN			
(b) $H^{4}$ , $H_{2}O$					

- Give the structural formula of the Organic Product. Circle the letter of the 5 reaction which you want counted. Write EC by the one reaction which you want counted as an extra credit problem If you do not choose, I will just grade the first 5 reactions and grade the 6<sup>th</sup> as extra credit.(EC worth 4 pts)(5 pts each, 15 pts)
  - 1) CH<sub>5</sub>-O-H + (CU<sub>3</sub>)<sub>3</sub>Si Cl -->
  - 2) CH<sub>3</sub>-O−II + CH<sub>3</sub>CH<sub>2</sub> CH<sub>2</sub> O → (show all possible products)

5) CH<sub>3</sub>-C=C-H <u>BH<sub>3</sub>/THF</u>, H<sub>2</sub>O<sub>2</sub>, H<sub>2</sub>O, NaOH

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Sign Name		
Please show work on all questions for par	rtial credit even on questions which do	not specify. I will only grade
legible answers because if I can't read yo	ur answer, I obviously cannot grade it.	NO EXCEPTIONS, (25 pts)
1. Keeping in mind the general react aldehyde, complete the following reaction you just give reaction products, you will !  CH <sub>3</sub> —C—CH <sub>3</sub> (a) O-CH <sub>2</sub> CH <sub>3</sub> (b) H <sup>+</sup> , H <sub>2</sub> O		

- 2. Give the structural formula of the Organic Product. <u>Circle the number of the 5 reaction which you want counted.</u> Write <u>EC</u> by the one reaction which you want counted as an extra credit problem If you do not choose, I will just grade the first 5 reactions and grade the 6<sup>th</sup> as extra credit.(EC worth 4 pts) (5 pts each,15 pts)
- f) CH<sub>3</sub>-O-H + CH<sub>3</sub> CH<sub>2</sub>-O-H <u>U''/ U<sub>2</sub>O</u> (show all possible products)
  - 2) CH<sub>3</sub>-O-CH<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub> + H Br (large excess) →